



Name _____ Class _____ Date _____

AFTER THIS WORKSHEET YOU CAN

- ✓ Understand: How a Heat Engine Works
- ✓ Understand: Efficiency of a Heat Engine
- ✓ Understand: The Carnot Engine
- ✓ Understand: Why Real Engines Are Less Efficient



Section A • Concept Check

• BEGINNER

4 × 1 = 4

- 1 A heat engine absorbs 2000 J and produces 600 J of work. Find its efficiency.

- 2 Find the Carnot efficiency between $T_h=800\text{K}$ and $T_c=400\text{K}$.

- 3 Find the Carnot efficiency between $T_h=1000\text{K}$ and $T_c=250\text{K}$.

- 4 Why does increasing the temperature difference (T_h-T_c) generally increase a heat engine's maximum possible efficiency?



Section B • Problem Solving

• INTERMEDIATE

4 × 3 = 12

- 5 A real engine has efficiency 25%, while its Carnot limit is 40%. What does this gap represent?

- 6 If a heat engine absorbs 1200 J and its efficiency is 35%, find the useful work output.

- 7 Using the previous engine, find how much heat is expelled to the cold reservoir.

- 8 Why must every cyclic heat engine expel some heat to a cold reservoir?



Section C • Challenge

• CHALLENGE

2 × 2 = 4

- 9 What happens to a heat engine's theoretical maximum efficiency as T_c approaches T_h ?

- 10 Why are power plants often built near rivers or large bodies of water?

Reflection — the most useful thing I learned: _____

✂ ANSWER KEY – FOLD OR CUT BEFORE HANDING OUT

1 = 0.3 or 30% 2 = $1 - (400/800) = 0.5$ or 50% 3 = $1 - (250/1000) = 0.75$ or 75% 4 = The ratio T_c/T_h decreases as the temperature difference grows, increasing $1 - (T_c/T_h)$ | 5 = Losses due to friction, heat leakage, and other non-ideal (irreversible) effects in the real engine 6 = $W = 0.35 \times 1200 = 420 \text{ J}$
7 = $Q_c = Q_h - W = 1200 - 420 = 780 \text{ J}$ 8 = The second law of thermodynamics prevents 100% conversion of heat into work in a repeating cycle | 9 = It approaches 0% efficiency, since the temperature difference driving the engine vanishes 10 = To provide a large, stable cold reservoir for expelling waste heat, improving the practical efficiency of the heat engine cycle